

Work each of the following problems. SHOW ALL WORK.

1. What are four similarities between electrostatic and gravitational forces? What are two differences?

Similarities:

1. Both forces decrease with the square of the distance between objects.

2. Both forces increase with a corresponding increase in charge or mass.

3. Both forces depend on fundamental physical constants: the gravitational constant for

gravitational force (G) and Coulomb's constant for electrostatic force (k).

4. Both forces act on a line between the objects involved,

either directly towards or directly away from one another.

5. Both forces are mediated by fields rather than by physical contact.

6. Both forces are additive, meaning the total force on one object is the sum of the forces

between that object and every other object.

Differences:

1. Electrostatic force can both attract and repel, while gravitational force can only attract.

2. Electrostatic force depends on charge, while gravitational force depends on mass.

3. All objects have a net mass and therefore experience gravitational force. However, not all

objects have a net charge and therefore might not experience electrostatic force.

2. Determine both the mass and the charge of a block of material consisting of 3 x 10²⁷ protons, 3 x 10³⁷ neutrons, and 3.1 x 10²⁷ electrons.

A proton's mass is 1.6×10^{-27} kg. A neutron's mass is also 1.6×10^{-27} kg.

An electron's mass is 9.1 x 10⁻³¹ kg. The total mass of the block is:

 $m_{block} = m_{p}$ (# of protons) + m_{p} (# of neutrons) + m_{e} (# of electrons)

$$\begin{split} m_{_{block}} &= (1.6 \times 10^{-27} \, kg) (3 \times 10^{27} \, protons) + (1.6 \times 10^{-27} \, kg) (3 \times 10^{27} \, neutrons) + (9.1 \times 10^{-31} \, kg) (3.1 \times 10^{27} \, electrons) \\ m_{_{block}} &= 9.6 \, kg \end{split}$$



Work each of the following problems. SHOW ALL WORK.

A proton's charge is 1.6 x 10⁻¹⁹ C. A neutron has no charge.

An electron's charge is -1.6 x 10⁻¹⁹ C. The total charge of the block is:

$$\begin{split} q_{block} &= q_{p} + q_{e} \\ q_{block} &= q_{p} (\# \ of \ protons) + q_{e} (\# \ of \ electrons) \\ q_{block} &= (1.6 \times 10^{-19} C) (3 \times 10^{27} protons) + (-1.6 \times 10^{-19} C) (3.1 \times 10^{27} electrons) \\ q_{block} &= -1.6 \times 10^{7} C \end{split}$$

3. A proton is placed 100 µm from a helium nucleus. The gravitational force pulls the proton and nucleus together, while the electric force pushes them apart. Which force is stronger and by how much?

A helium nucleus has two protons. The charge of a single proton is 1.6 x 10⁻¹⁹ C,

and its mass is 1.6 x 10⁻²⁷ kg. A neutron's mass is equal to a proton's mass.

The electric force pushing the proton and nucleus apart is:

$$F_{electric} = \frac{k |q_{proton} q_{nucleus}|}{r^2}$$

$$F_{electric} = \frac{(9 \times 10^9 Nm^2/C^2)(1.6 \times 10^{-19} C)(2)(1.6 \times 10^{-19} C)}{(1 \times 10^{-4} m)^2}$$

$$F_{electric} = 4.6 \times 10^{-20} N$$

The gravitational force pulling the proton and nucleus together is:

$$F_{gravity} = \frac{Gm_{proton} m_{nucleus}}{r^2}$$

$$F_{gravity} = \frac{(6.6 \times 10^{-11} m^3/kgs^2)(1.6 \times 10^{-27} kg)(4)(1.6 \times 10^{-27} kg)}{(1 \times 10^{-4} m)^2}$$

$$F_{gravity} = 1.7 \times 10^{-56} N$$

$$\frac{F_{electric}}{F_{gravity}} = \frac{4.6 \times 10^{-20} N}{1.7 \times 10^{-56} N} = 2.7 \times 10^{35}$$

The electric force is stronger by roughly a trillion trillion times!



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4. How far apart are a proton and an electron if they exert an attractive force of 3 N on each other?

A proton's charge is 1.6×10^{-19} C and an electron's charge is -1.6×10^{-19} C.

The electric charge between them is:

 $F_{electric} = \frac{k|q_{proton} q_{electron}|}{r^2}$

Rearranging for r, the distance between the charges is:

$$r = \sqrt{\frac{kq_{proton} |q_{electron}|}{F_{electric}}}$$
$$r = \sqrt{\frac{(9 \times 10^9 Nm^2/C^2)(1.6 \times 10^{-19} \text{ C})(|-1.6 \times 10^{-19} \text{ C}|)}{3 N}}$$
$$r = 8.8 \times 10^{-15} m$$

- 5. If the total charge of an atom's nucleus is +3 and the total charge of the surrounding electrons is -3, the atom is which one of the following:
 - a. positively charged
 - b. negatively charged
 - c. electrically neutral
 - d. unstable

c. electrically neutral



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6. An object with charge 4.3 x 10⁻⁵ C pushes another object 0.31 μ m away with a force of 7 N. What is the total charge of the second object?

Refer to Coulomb's law:

$$F_{electric} = \frac{k |q_1 q_2|}{r^2}$$

 $q_{2} = \frac{F_{electric} r^{2}}{kq_{1}}$ $q_{2} = \frac{(7 N)(3.1 \times 10^{-7} m)^{2}}{(9 \times 10^{9} Nm^{2}/C^{2})(4.3 \times 10^{-5} C)}$ $q_{2} = 1.7 \times 10^{-18} C$

- 7. A balloon, which is initially neutral, is rubbed with fur until it acquires a net charge of -0.40 nC.
 - a. Assuming that only electrons are transferred, were electrons removed from the balloon or added to it?

The balloon started neutral and ended with a net negative charge,

indicating that electrons were transferred to it.

b. How many electrons were transferred?

The charge of an electron is -1.6 x 10⁻¹⁹ C.

of electrons transferred = $\frac{-0.40 \times 10^{-9} \text{ C}}{-1.6 \times 10^{-19} \text{ C}}$ = 2.5 × 10° or two and a half billion electrons



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8. Two +1 C charges are separated by 3,000 m. What is the magnitude of the electric force between them?

Refer to Coulomb's law:
$$F_{electric} = \frac{k|q_1q_2|}{r^2}$$
$$q_1 = 1C$$
$$q_2 = 1C$$
$$r = 3000 m$$
$$F_{electric} = \frac{9 \times 10^9 Nm^2/C^2(1C)(1C)}{(3000 m)^2}$$
$$F_{electric} = 1000 N$$